Atmospheric and Oceanic Sciences 3/3L

Introduction to the Atmospheric Environment

Instructor Dr. Jeffrey Lew

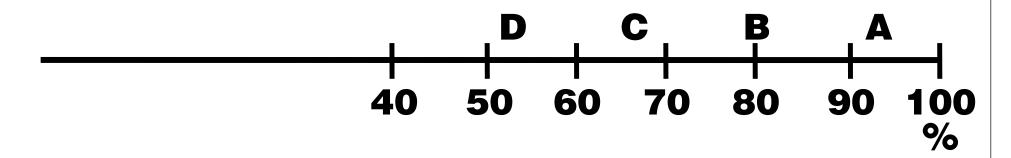
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Grades

- 2 closed-book exams, 650 points total
- 3 take-home quizzes, 150 points total
- 5 of 7 in-lecture quizzes, 100 points total
- 4 in-discussion quizzes, 100 points total



Laboratory Section (3L)

GE requirement: Foundations of Scientific Inquiry—2 courses from Physical Sciences, one of which is a 5-unit course with lab/demo or Writing II credit.

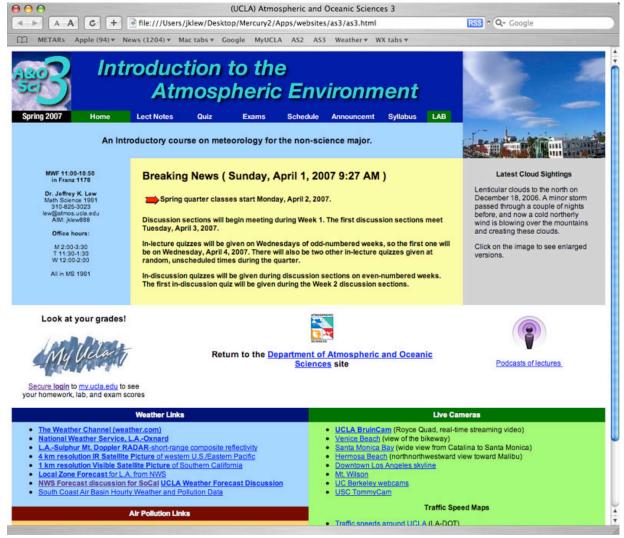
Lecture/discussion counts as a 4-unit course

Take 1-unit lab at same time to fulfill 5-unit requirement, if needed

In any case, lab can only be taken concurrently with lecture/discussion

Web Site

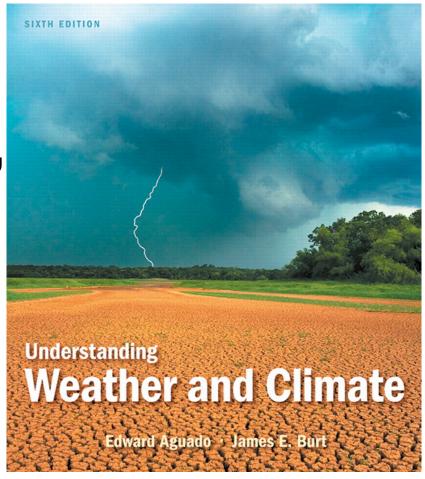
www.atmos.ucla.edu/AS3/as3.html



my.ucla.edu

Textbook

E. Aguado and J.E. Burt, *Understanding Weather and Climate*, 6th ed.





Podcasts

Audio recordings of lectures, enhanced with slide builds



Go to class web site to get instructions on how to subscribe using iTunes

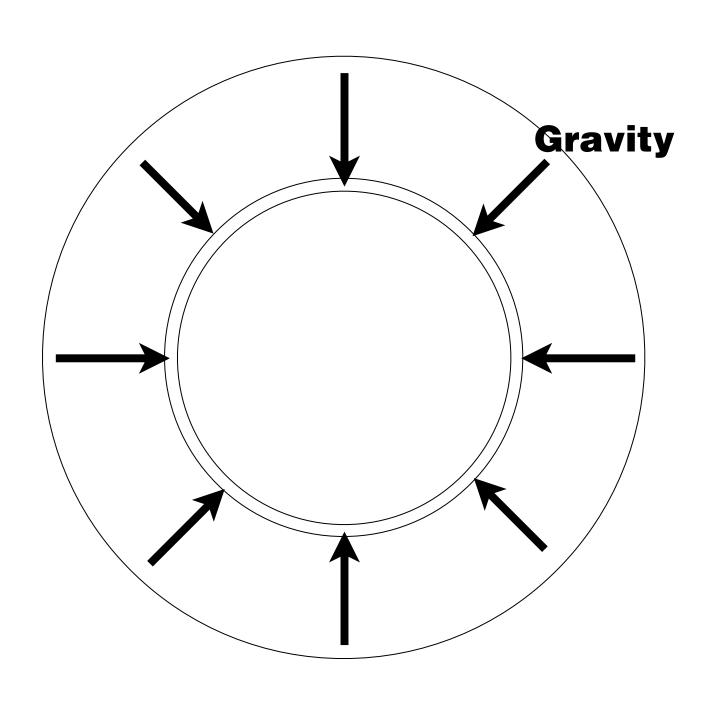
Ch. 1: Overview of the Atmosphere and Weather

The Atmosphere

- → Gaseous Composition
- → Vertical Density, Temperature, and Pressure Profiles
- → Atmospheric Layers
- → Evolution of Earth's Atmosphere

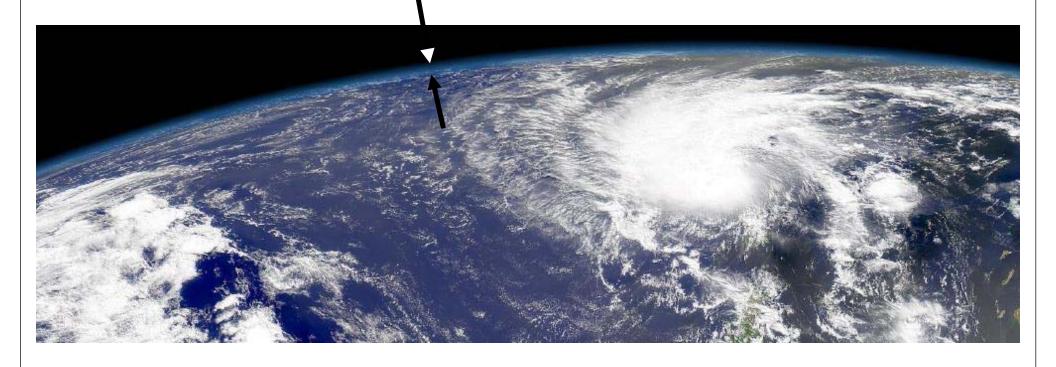
Weather

- → Definitions and Disciplines of Study
- → Weather Elements
- → Historical Highlights



Atmospheric Dimensions

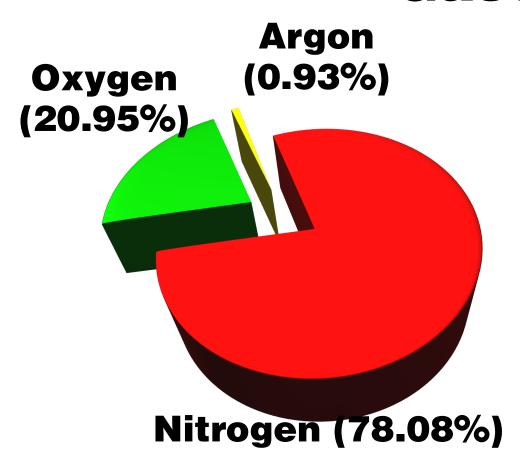
100-500 km



Total mass: 5×10^{18} kg

Atmospheric Composition

Permanent/Non-variable/Fixed Gases



Permanent Trace Gases

> Neon Helium Hydrogen Xenon Krypton

TABLE 1-2 Permanent Gases of the Atmosphere

Constituent	Formula	Percent by Volume	Molecular Weight
Nitrogen	N_2	78.08	28.01
Oxygen	O_2	20.95	32.00
Argon	Ar	0.93	39.95
Neon	Ne	0.002	20.18
Helium	He	0.0005	4.00
Krypton	Kr	0.0001	83.8
Xenon	Xe	0.00009	131.3
Hydrogen	H ₂	0.00005	2.02

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Variable Gases

 Thousands of gases, whose concentrations vary over short time scales:

→ Water Vapor (H₂O)

- 1–4%, depending on temperature
- Mostly located below 10 km altitude

→ Carbon Dioxide (CO₂)

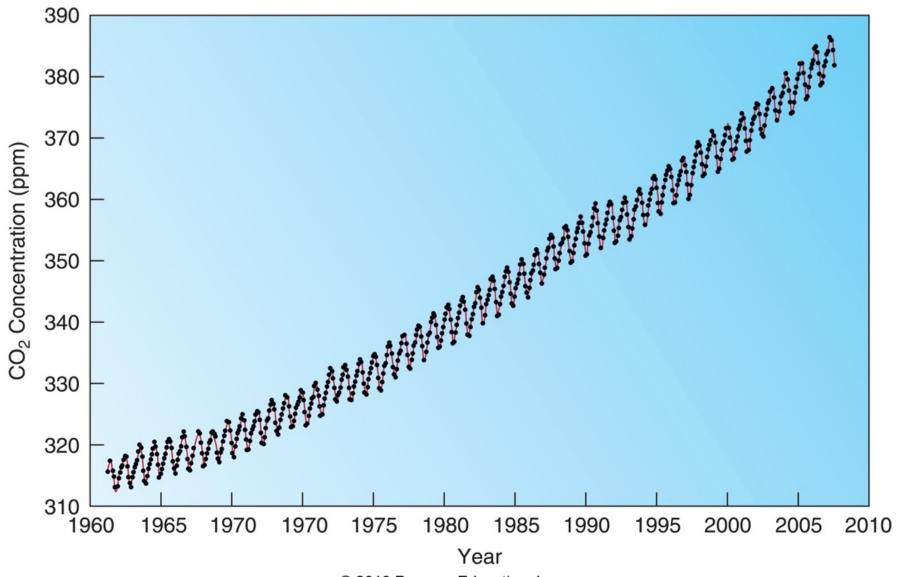
0.038% ±0.0006%, depending on season

→ Ozone (O₃)

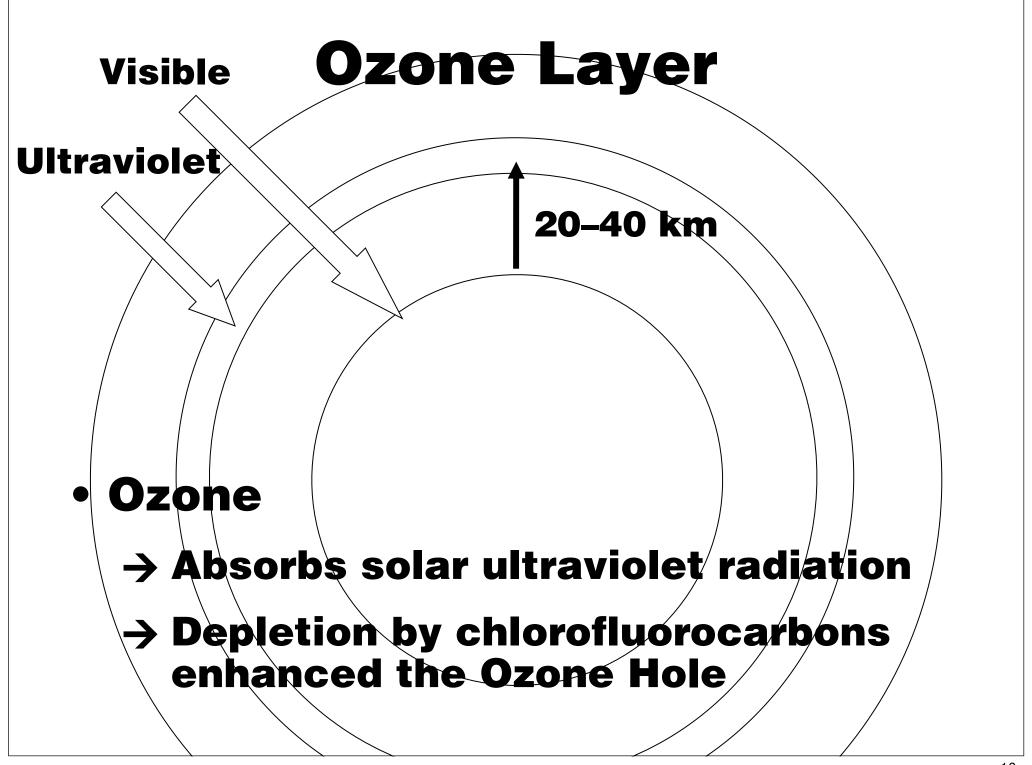
 Concentration varies with location (urban smog, stratospheric ozone layer)

Important Features of Variable Gases

- Water Vapor
 - → Heat transport
 - → Hydrologic Cycle
- Carbon Dioxide
 - → Greenhouse Effect
 - → Respiration/Green Plant Photosynthesis
 - → Increasing due to human activities

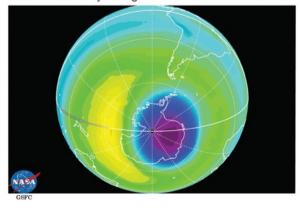


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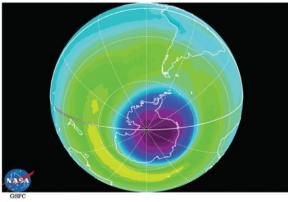


Focus 1-2 (pp. 12-13)

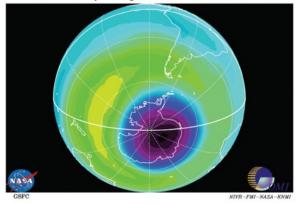
Nimbus-7 Monthly Average Total Ozone for October 1992



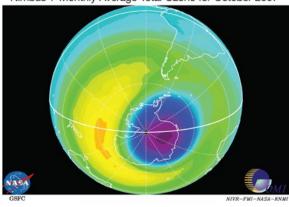
Nimbus-7 Monthly Average Total Ozone for October 1997

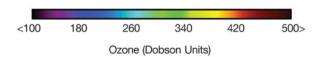


Nimbus-7 Monthly Average Total Ozone for October 2006



Nimbus-7 Monthly Average Total Ozone for October 2007



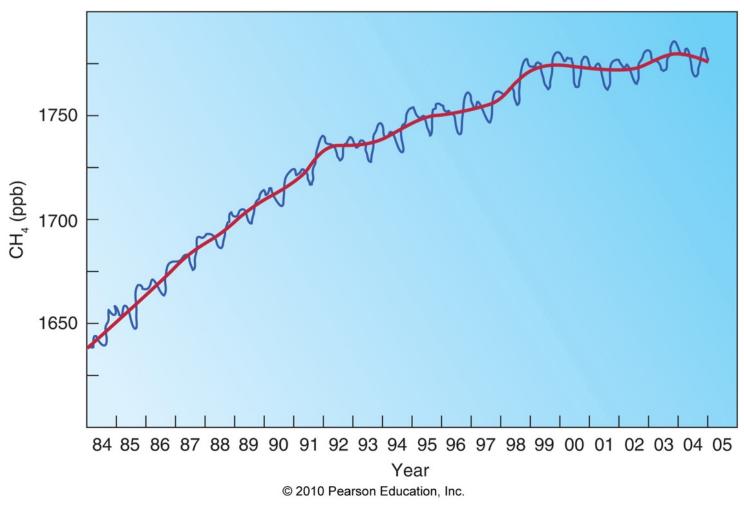


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Methane

→ From organic decay, fossil fuel leaks

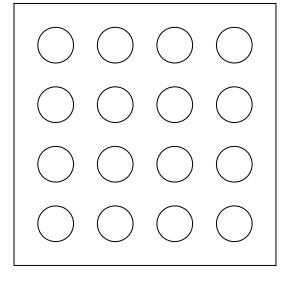
→ Also a greenhouse gas

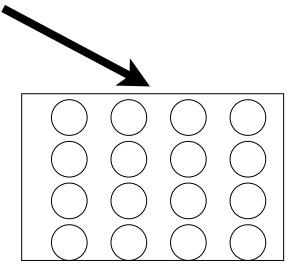


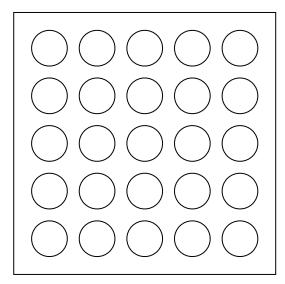
Physical Structure of the Atmosphere

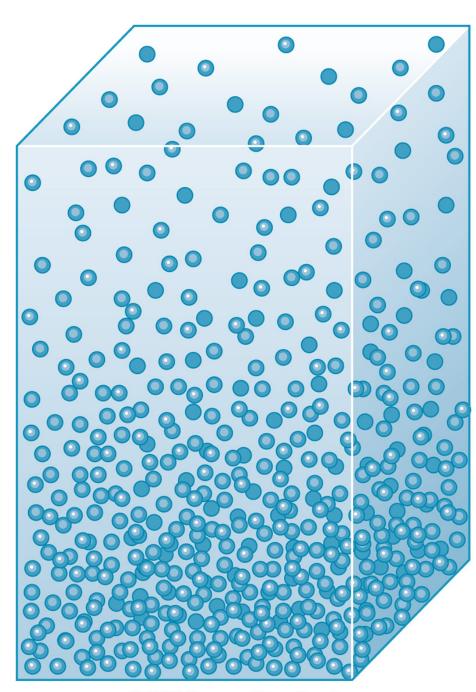
Density: mass per unit volume

$$\rho = \frac{\mathbf{M}}{\mathbf{V}}$$

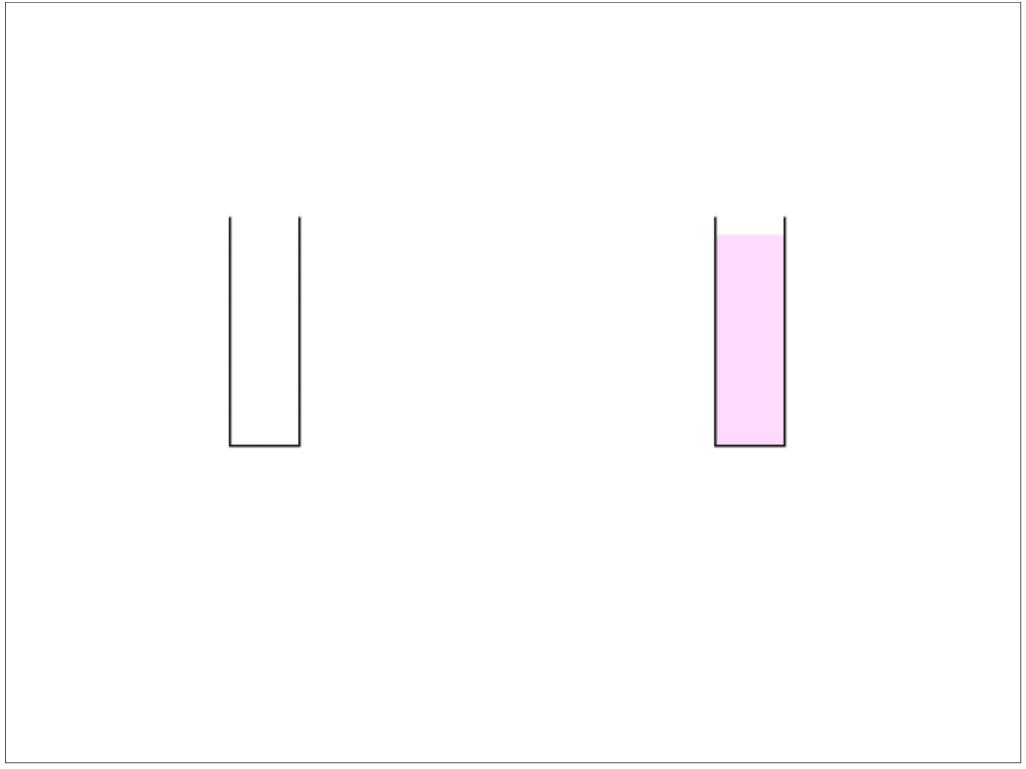








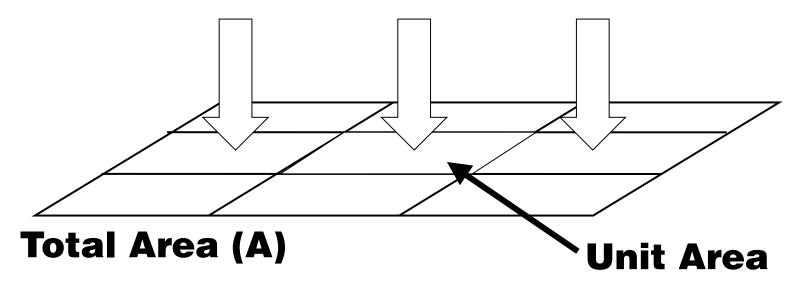
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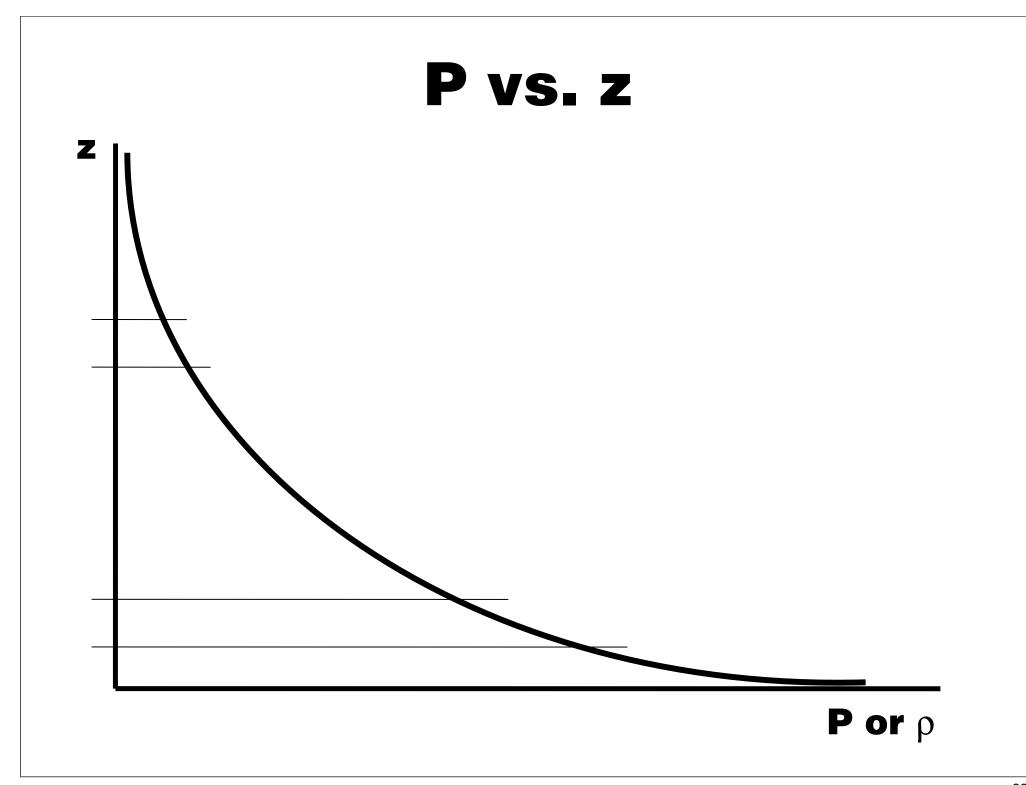


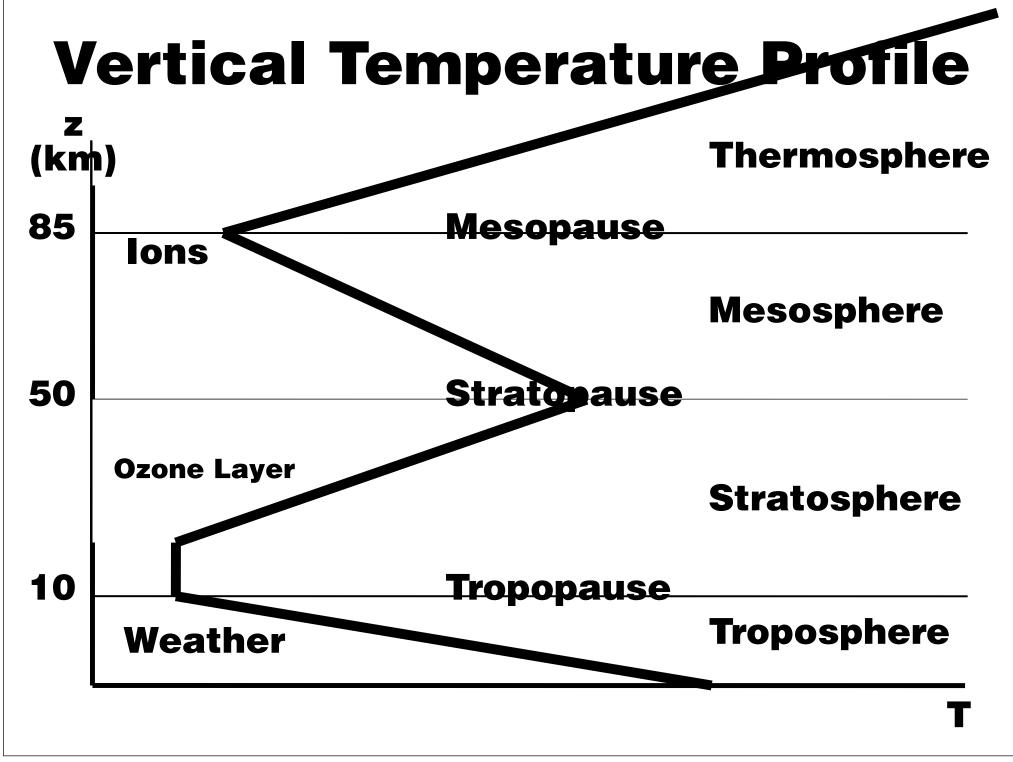
Pressure = Force per Unit Area

$$P = \frac{F}{A}$$

Force (F)

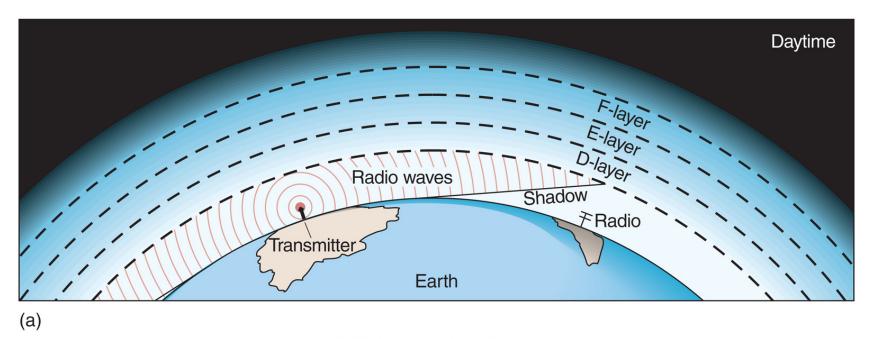




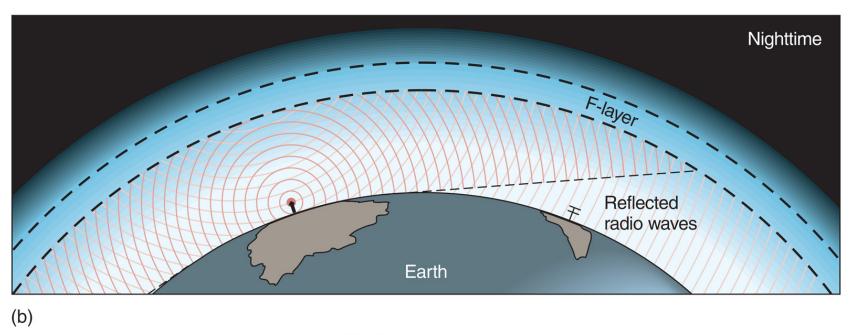


Ionosphere

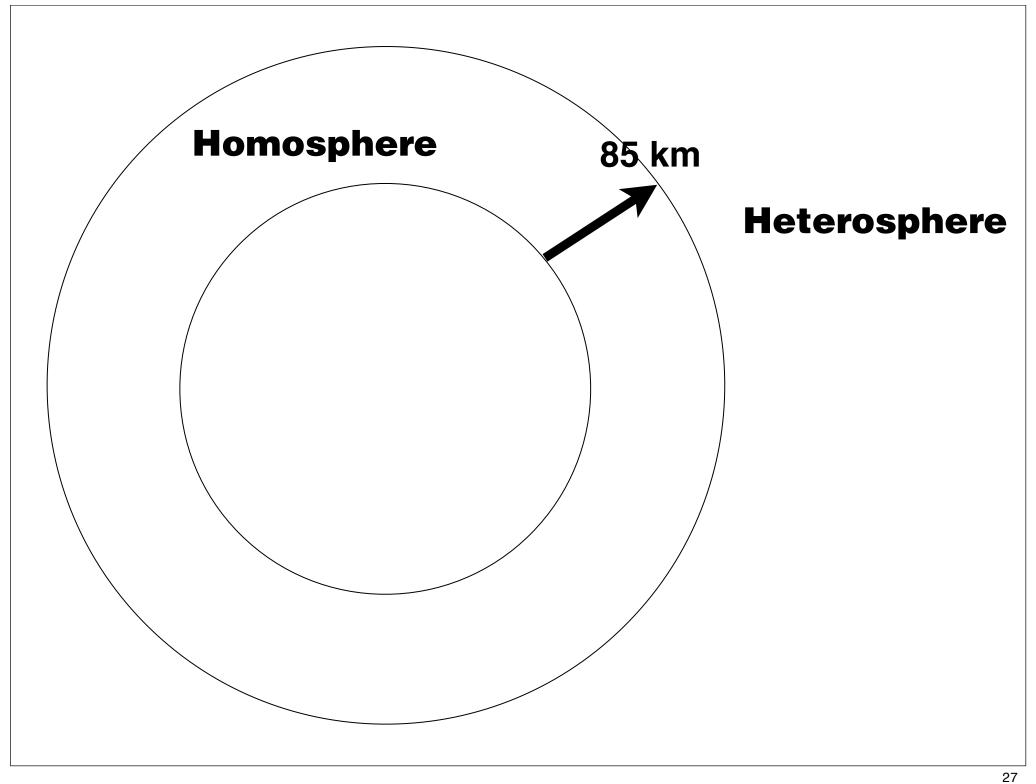
- Outer layers of atmosphere exposed to strong sunlight
 - → Produces electrically charged ions



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Atmospheric Evolution

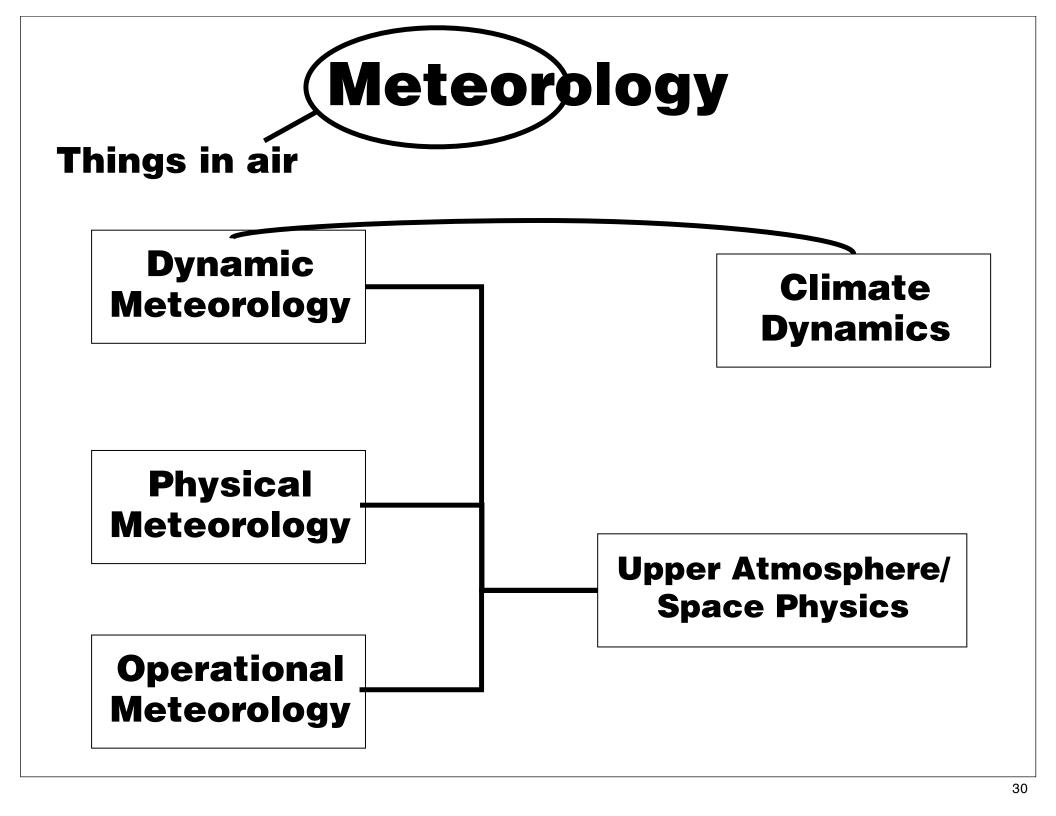
- Primordial Atmosphere: 4.5 BYA
 - → Condensation of Interstellar Matter (mostly Hydrogen and Helium)

- Secondary Atmosphere: 4 BYA
 - → Formed by planetary outgassing (mostly Water Vapor and Carbon Dioxide)
 - → Later, water vapor condensed out to form oceans, and carbon dioxide dissolved into the ocean water

- Green "Plants": 2.5 BYA
 - → Green Plant Photosynthesis released molecular oxygen into the environment
- Present Atmosphere
 - → Lifeforms and atmosphere co-evolved to form "Class-M" environment: nitrogen/ oxygen



Cyanobacter ia



Definitions

- Weather
 - → An observation of the weather elements at one point in time
- Climate
 - → A summary of a set of weather observations taken over a period of time

Weather Elements

Temperature





Pressure

Humidity





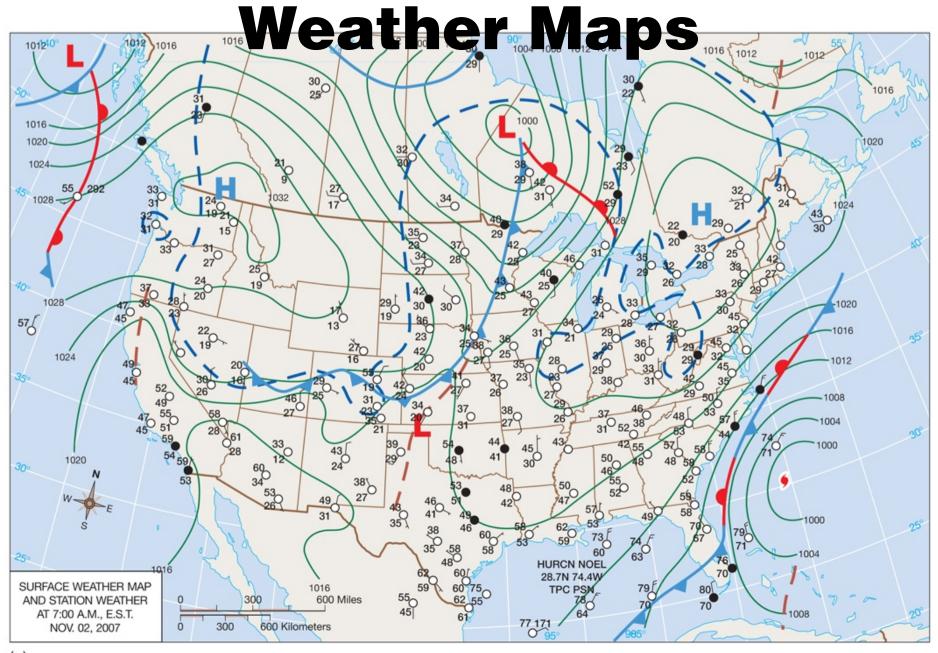
Wind





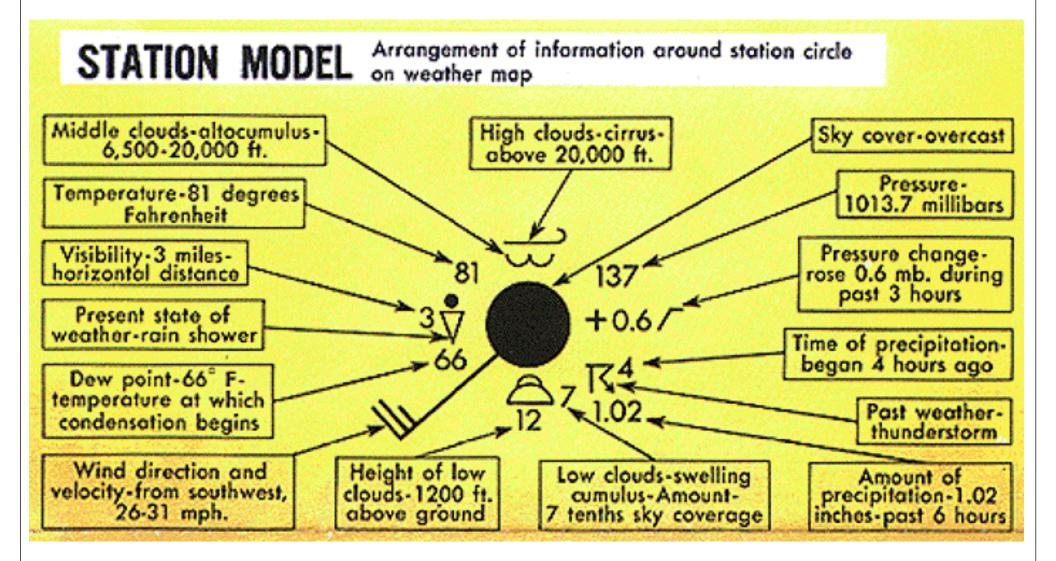


Precipitation



(a)

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Historical Highlights

Babylonians, Chinese: Astrometeorology

34 BC: Aristotle, Father of Meteorology

Bacon/Descartes—Scientific Method

Instrumentation

Telegraph

Computers, satellites

21st c.: Modern Meteorology